

the permissible energy levels of matter are quantized.

idea of intro to this lecture

→ distinguish materials from one another.

→ think "closed shells" like magic #s

→ gives differences in physical properties (e.g. transparency, conductivity)

→ and size-dependent properties @ nanoscale.

by interactions among energy carriers

how was this developed?

19th century: newtonian = particles

electromagnetism = waves.

changes rate to energy of charged particles.

⇒ discrete lines in absorption spectra of line gases. e.g. (12 characteristic)

planck: $E_p = h \nu$ } frequency = $n h \nu$ } quantum #

$h = 6.6 \times 10^{-34} \text{ J}\cdot\text{s}$

"discrete energy" ⇒ later called a photon.

this was used to explain the photoelectric experiment,

$h \nu \geq E_v - E_f$ } (put on slide w/p.e. effect)

work function = minimum binding energy

Einstein $E = mc^2$, photon has a mining mass.

therefore, minimum $p = mc = \frac{E_f}{c} = \frac{h \nu}{c} = \frac{h}{\lambda} = \frac{\hbar}{k}$

frequency / wavelength. $\hbar = \frac{h}{2\pi}$

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now we can relate energy + momentum to frequency and wavelength.

wave/particle duality: "double slit" experiment, backdrop is perfect absorber.

- "bullet" experiment, $P_1 + P_2 = P_{12}$

- "water pebble" experiment, interference

$$I_{12} = I_1 + I_2 + 2\sqrt{I_1 I_2} \cos \phi$$

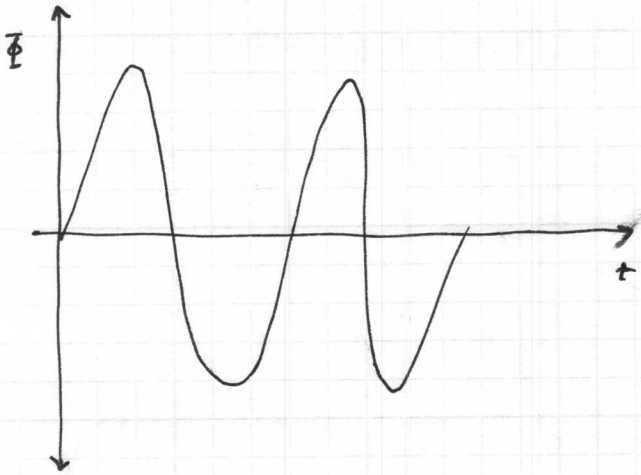
phase based on point on "detector"

- electrons: arrive one at a time, yet interfere like waves.

electron is simultaneously interacting w/ both slits!

uncertainty principle: cannot observe w/o interacting

now let's consider a simple wave, in 1D.



$$\Phi = A \sin \left(2\pi \nu t - \frac{2\pi x}{\lambda_x} \right) \hat{y} = A \sin (\omega t - k_x x) \hat{y}$$

has periodicity in space and time

this is a traveling wave

- sit stationary in space, at a point, and we see an oscillation in time
- sit stationary in time, and see an oscillation in space

$\omega = 2\pi\nu$, $\omega = \text{angular frequency}$

wavevector $k = \frac{2\pi}{\lambda_x} \hat{x} = k_x \hat{x}$
 ↖ magnitude of wavevector.

how fast is the wave moving?

constant phase plane = $\omega t - kx = \text{constant}$

$$\frac{dt}{dx} = \frac{\omega}{k} = \nu \lambda = v = \underline{c}$$

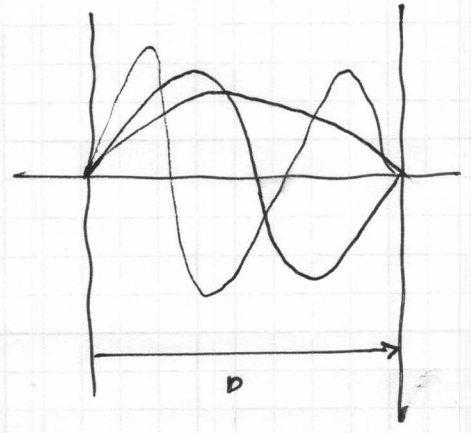
speed of light.

now consider a wave "pinned" at its ends, a standing wave
add a left-moving wave to a right-moving wave

$$\underline{\Phi} = (A \sin(\omega t - kx) + A \sin(\omega t + kx)) \hat{y}$$

↑ $k_x = k$ (1D system)

$$= -2A \cos(\omega t) \sin(kx) \hat{y}$$



Boundary conditions: $\underline{\Phi}(0) = \underline{\Phi}(D) = 0$

therefore, $\sin(kx) = 0$, because $\cos(\omega t) \neq 0$ unless $t = \frac{n\pi}{2}$

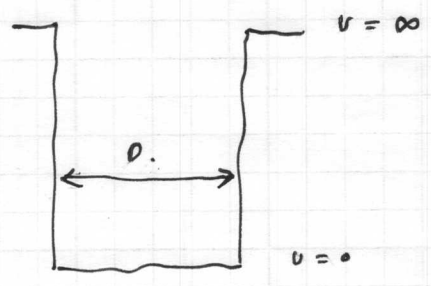
$$\sin\left(\frac{2\pi}{\lambda} x\right) = 0$$

solved if $D - \frac{2\pi}{\lambda} = n\pi$ ~~==>~~ ~~sketch~~

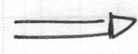
$$\Rightarrow D = \frac{n\lambda}{2}$$

1D constraint for a standing wave "pinned" at its ends.

if this is a "particle in a box" (infinite potential well)



we know $E = Mc^2$



sketch
later
w/ schrodinger
eqn.

relating momentum and energy, as before,

$$p = \frac{h}{\lambda} = \frac{h\nu}{2D}$$

$$E = mc^2 = \frac{p^2}{2m} = \frac{1}{2m} \left(\frac{h^2 \nu^2}{4D^2} \right) = \frac{1}{2m} \left(\frac{h\nu}{2D} \right)^2$$

we formalize this in QM using Schrodinger's eqn, which applies to the wavefunction of any matter

(ψ)

$$-\frac{\hbar^2}{2m} \nabla^2 \psi + V\psi = i\hbar \frac{d\psi}{dt}, \quad \hbar = \frac{h}{2\pi}, \quad V = \text{potential energy}$$

function of position and time

wavefunction: probability that matter is at position r at time t
 ↑
 density function

$$\text{so } \int_{-\infty}^{\infty} \psi \psi^* dx = 1$$

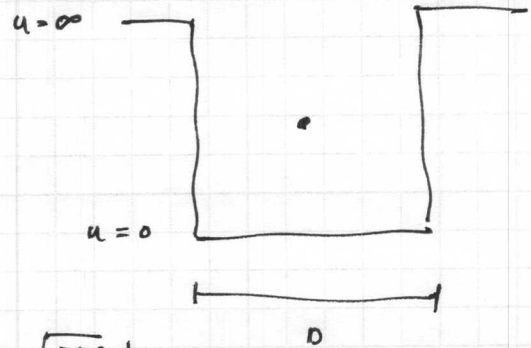
steady-state Schrodinger eqn, when $\psi(r,t) = \psi(r)$, varies with position only
 separate variables to give

$$-\frac{\hbar^2}{2m} \nabla^2 \psi + (V-E)\psi = 0$$

↑
 eigenvalues
 = allowed energy states.

now consider particle in a 1D well, width D .

$$-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} + (V-E)\psi = 0$$



general solution

$$\psi = A \exp\left\{-ix\sqrt{\frac{2mE}{\hbar^2}}\right\} + B \exp\left\{ix\sqrt{\frac{2mE}{\hbar^2}}\right\}$$

for $0 < x < D$

$\psi = 0$ elsewhere.

BC's.

$$\text{at } x=0: A+B=0$$

$$\text{at } x=D: A \exp\left\{-iD\sqrt{\frac{2mE}{\hbar^2}}\right\} + B \exp\left\{iD\sqrt{\frac{2mE}{\hbar^2}}\right\} = 0$$

$$\text{solve this system: } \sin\left(D\sqrt{\frac{2mE}{\hbar^2}}\right) = 0 \quad \sin(n\pi) = 0$$

$$\Rightarrow n\pi = D\sqrt{\frac{2mE}{\hbar^2}}$$

$$\Rightarrow E_n = \frac{1}{2m} \left(\frac{\hbar n}{2D}\right)^2$$

if we solve for a 2D case, we have two quantum numbers.

$$E_{2n} = \frac{(l^2 + n^2) \pi^2 \hbar^2}{2mD^2}$$

harmonic oscillator: model for a single bond.

general shape of interatomic potential. } — see slide: graphic (chem)

$$v(x') = v(x_0) + \frac{1}{2} \left[\frac{d^2 v}{dx'^2} \right]_{x'=x_0} (x' - x_0)^2 + o(x' - x_0)^3 + \dots$$

$$F = -\frac{dv}{dx'} = -k(x' - x_0)$$

↑ spring constant, so $v = \frac{kx'^2}{2}$

$$-\frac{\hbar^2}{2m} \left(\frac{d^2 \psi}{dx'^2} \right) + \left(\frac{kx'^2}{2} - E \right) \psi = 0$$

$$\psi(0) = \text{finite}$$

$$\psi(x \rightarrow \infty, x \rightarrow -\infty) = 0$$

solve to get $E_n = \left(n + \frac{1}{2} \right) \hbar \nu$

$$\nu = \frac{1}{2\pi} \sqrt{\frac{k}{m}}$$

↑ natural frequency!

$$\text{mass} = \frac{m_1 m_2}{m_1 + m_2}$$

⇒ energy is distributed in different modes

⇒ connection to optical spectroscopy which measures bond resonances.